

## **BASIC RULES FOR UNCERTAINTY CALCULATIONS**

### Types of uncertainties:

- 1) An Absolute Uncertainty is denoted by the symbol “ $\Delta$ ” and has the same units as the quantity.
- 2) A Relative or Percent Uncertainty is denoted by the symbol “ $\epsilon$ ” and has no units.

To convert back and forth between the two types of uncertainties consider the following:

$$m = (3.3 \pm 0.2) \text{ kg} = (3.3 \text{ kg} \pm 6.1\%)$$

The Absolute Uncertainty is:  $\Delta m = 0.2 \text{ kg} = (6.1/100) \times 3.3 \text{ kg}$

The Relative Uncertainty is:  $\epsilon_m = 6.1\% = (0.2/3.3) \times 100\%$

### Completing Uncertainty Calculations:

- 1) Addition and Subtraction: ADD the Absolute Uncertainties

**Rule:**  $(A \pm \Delta A) + (B \pm \Delta B) = (A + B) \pm (\Delta A + \Delta B)$   
 $(A \pm \Delta A) - (B \pm \Delta B) = (A - B) \pm (\Delta A + \Delta B)$

Consider the numbers:  $(6.5 \pm 0.5) \text{ m}$  and  $(3.3 \pm 0.1) \text{ m}$

Add:  $(6.5 \pm 0.5) \text{ m} + (3.3 \pm 0.1) \text{ m} = (9.8 \pm 0.6) \text{ m}$

Subtract:  $(6.5 \pm 0.5) \text{ m} - (3.3 \pm 0.1) \text{ m} = (3.2 \pm 0.6) \text{ m}$

- 2) Multiplication and Division: ADD the Relative Uncertainties

**Rule:**  $(A \pm \epsilon_A) \times (B \pm \epsilon_B) = (A \times B) \pm (\epsilon_A + \epsilon_B)$   
 $(A \pm \epsilon_A) / (B \pm \epsilon_B) = (A / B) \pm (\epsilon_A + \epsilon_B)$

Consider the numbers:  $(5.0 \text{ m} \pm 4.0\%)$  and  $(3.0 \text{ s} \pm 3.3\%)$

Multiply:  $(5.0 \text{ m} \pm 4.0\%) \times (3.0 \text{ s} \pm 3.3\%) = (15.0 \text{ m} \cdot \text{s} \pm 7.3\%)$

Divide:  $(5.0 \text{ m} \pm 4.0\%) / (3.0 \text{ s} \pm 3.3\%) = (1.7 \text{ m/s} \pm 7.3\%)$

- 3) For a number raised to a power, fractional or not, the rule is simply to MULTIPLY the Relative Uncertainty by the power.

**Rule:**  $(A \pm \epsilon_A)^n = (A^n \pm n\epsilon_A)$

Consider the number:  $(2.0 \text{ m} \pm 1.0\%)$

Cube:  $(2.0 \text{ m} \pm 1.0\%)^3 = (8.0 \text{ m}^3 \pm 3.0\%)$

Square Root:  $(2.0 \text{ m} \pm 1.0\%)^{1/2} = (1.4 \text{ m}^{1/2} \pm 0.5\%)$

- 4) For multiplying a number by a constant there are two different rules depending on which type of uncertainty you are working with at the time.

**Rule - Absolute Uncertainty:**  $c(A \pm \Delta A) = cA \pm c(\Delta A)$

Consider:  $1.5(2.0 \pm 0.2) \text{ m} = (3.0 \pm 0.3) \text{ m}$

Note that the Absolute Uncertainty **is** multiplied by the constant.

**Rule - Relative Uncertainty:**  $c(A \pm \epsilon_A) = cA \pm \epsilon_A$

Consider:  $1.5(2.0 \text{ m} \pm 1.0\%) = (3.0 \text{ m} \pm 1.0\%)$

Note that the Relative Uncertainty **is not** multiplied by the constant.

### Comparing Values:

- 1) When comparing two values you should complete a CONSISTENCY CHECK.  
This is the preferred way to compare values which have uncertainties.

**Rule:** If the following inequality is true you may say that the values you are comparing are consistent with each other within experimental uncertainty, otherwise the values are inconsistent.

$$|A - B| \leq |\Delta A + \Delta B|$$

Consider the numbers:  $(3.3 \pm 0.2) \text{ m}$  and  $(3.1 \pm 0.1) \text{ m}$

Consistency check:  $|3.3 - 3.1| \text{ m} \leq |0.2 + 0.1| \text{ m}$   
 $|0.2| \text{ m} \leq |0.3| \text{ m} \quad \therefore \text{Consistent Values}$

Note that the difference of the two quantities is less than or equal to the sum of the Absolute Uncertainties. You cannot complete a consistency check using Relative Uncertainties.

Sometimes you will be comparing an experimental value which has an uncertainty to a theoretical value which may not have an uncertainty. In this case the above calculation still holds.

Compare the numbers: $g = 9.87 \text{ m/s}^2 \pm 0.09 \text{ m/s}^2$	Experimental value
$g = 9.81 \text{ m/s}^2$	Theoretical Value
Consistency check: $ 9.87 - 9.81  \text{ m/s}^2 \leq  0.09  \text{ m/s}^2$	
$ 0.06  \text{ m/s}^2 \leq  0.09  \text{ m/s}^2$	Consistent

- 2) When comparing two values which don't have uncertainties you may then calculate a Percent Difference as follows:

$$\% \text{ Diff} = \frac{|Theor.Value - Exp.Value|}{Theor.Vlaue} \times 100\%$$